

"Deposition of Colloidal Titanium Oxide Nanoparticles using Pulsed Nd: YAG Laser Ablation in Liquid to Increase the Efficiency of the Si Solar Cell"

Amenah Ali Salman, Hiba S. Tarik and Khaled Z. Yahea

Department of Applied Science, University of Technology, Baghdad-Iraq.

Abstract

This work reports the attempts to carry out pulsed laser ablation in liquid (PLAL) for synthesizing colloidal Titanium dioxide nanoparticles (NPs). TiO₂ NPs was synthesized by 7ns Nd:YAG laser ablation of high purity titanium target (99.99 %) immersed in water, and deposited the film on silicon solar cell to increase the efficiency of the Si solar cell. The surface morphology of the deposits materials have been studied by using atomic force microscopes (AFM). Atomic Force Microscopy (AFM) analysis showed that the average grain size of TiO₂ NPs in water were 150 nm and rms roughness values are (2.27 nm) for TiO₂ thin film. The photovoltaic characteristics before and after deposited Titanium dioxide nanoparticles (NPs) included short circuit current (J_{sc}), open circuit voltage (V_{oc}), where the maximum (J_{sc}), (V_{oc}) and fill factor (FF) obtained at AM1 after deposited were 40.2 (mA cm⁻²), 630(mV) and (0.7) respectively. After deposited Titanium dioxide nanoparticles (NPs) increase the efficiency of the Si solar cell 13.8 % instead of 10.3% conversion efficiency before the deposited.

Keywords: Laser ablation, Titanium dioxide nanoparticles in liquids, solar cell.

Introduction

TiO₂ has been investigated for many years. Titanium dioxide is a fascinating class of inorganic solids in a wide range of common and high technique applications due to its wide application in the photocatalysis, optical materials, dye-sensitized solar cell, and lithium-ion batteries fields [1–4]. Crystalline TiO₂ exists in three forms: rutile, anatase and brookite, of which, rutile is the most thermodynamically stable phase [5]. In recent years, pulsed laser ablation of metal target in liquid media has attracted great interest because such laser ablation in liquid (LAL) can produce the extreme conditions and lead to the formation of the novel nanostructures [6–10]. When a pulsed laser beam with enough energy irradiates on a metal target in transparent liquid, a local plasma with super-high temperature (about 6000 K) and high-pressure (about 1 GPa) will instantly be produced on the solid-liquid interface, and quench quickly after one pulse due to adiabatic expansion of the plasma and its interaction with surrounding media. The whole process is finished in about 1 μs. So, the laser ablation of metal targets in liquid media can form some special nanomaterials that are difficult to be obtained by the conventional methods [11–14].

In this work, we reported successful one-step fabrication of rutile TiO₂ nanoparticles by

LAL of titanium plate in distilled deionized water at room temperature.

In this work, we have performed laser ablation of Titanium target in water to synthesize Titanium oxide nanoparticles and deposited the film on silicon solar cell to increase the efficiency of the Si solar cell. The surface morphology of the deposits materials have been studied by using atomic force microscopes (AFM).

Experiment

Titanium oxide NPs were produced by laser ablation of high purity Titanium target immersed in water at room temperature. Fig.(1) displays the schematic diagram of experimental set-up of PLAL system. The Titanium target is placed in the bottom of glasses vessel filled with 2ml of liquid. The Titanium target was irradiated with Q-switched Nd:YAG laser operated at wavelength of 1064nm, 7ns pulse duration, and repetition frequency of 1Hz. The laser energy was used to ablate Titanium target was 80mJ/pulse, the ablation time was 5min. The laser beam was focused on Titanium target using focusing lens of 50mm focal length. Ablated Ti NPs were analyzed by atomic force microscopy (AFM) model (SPMAA 3000 Angstrom Advanced Inc. USA).

Result and Discussion

After laser ablation of Titanium target, the color of suspension is changed from colorless to white color as depicted in Fig.(2), indicating the production of Titanium oxide colloidal nanoparticles [11].

Fig.(3) demonstrate 3D atomic force microscopes AFM images of Titanium oxide NPs ablated in water with scanning area of $1\mu\text{m}\times 1\mu\text{m}$. The average grain size reaches 81.04 nm in diameter and its height around 7nm. Andrms roughness values are (2.27 nm) for TiO_2 thin film. It can be noticed from AFM images that particles have different morphologies with nanoparticles sizes.

Fig.(4) shows the short circuit current density J_{sc} as a function of illuminating power before and after deposited Titanium oxide NPs, at low levels of illuminating powers we noted that the J_{sc} have a linearity behavior with increasing power. But at high levels of illuminating power J_{sc} have a exponentially behavior that explained to the saturate in carriers, the maximum (J_{sc}) after deposited Titanium oxide NP were $40.2 \text{ (mA cm}^{-2}\text{)}$.

Fig.(5) shows the open circuit voltage V_{oc} as a function of illuminating power before and after deposited Titanium oxide NPs, we noted that the V_{oc} increase when illuminating power increase, the maximum (V_{oc}) after deposited Titanium oxide NPs were 630(mV).

The photovoltaic performance is shown in Fig.(6.a,b) in which the power can be extract from the cell before and after deposited Titanium oxide NPs from this curve we obtained the open circuit voltage (V_{oc}) is 630mV while short circuit current density (J_{sc}) is 40.2 mA/cm^2 and fill factor ($FF=0.7$). The high fill factor is probably due to high shunt resistance. The higher short circuit current density may be because the photons is due to carriers that are generated deep in the bulk of the silicon.

Fig.(7 a,b) demonstrates the variation of the output power (the power generated by the cell under simulated (AM1) versus voltage across the load resistance before and after deposited Titanium oxide NPs, After deposited Titanium dioxide nanoparticles (NPs) increase the efficiency of the Si solar cell 13.8 %

instead of 10.3% conversion efficiency before the deposited.

Conclusion

Colloidal Titanium dioxide nanoparticles have been prepared successfully by pulsed laser ablation of Titanium target in water. Formation of Titanium dioxide nanoparticles was emphasizing by atomic force microscopy (AFM) confirmed that synthesized TiO_2 NPs have different grain size and morphology showed that the average grain size of TiO_2 NPs in water were 81.04 nm and rms roughness values are (2.27 nm). The photovoltaic characteristics after deposited Titanium dioxide nanoparticles (NPs) included short circuit current (J_{sc}), open circuit voltage (V_{oc}), where the maximum (J_{sc}), (V_{oc}) and fill factor (FF) obtained at AM1 were $40.2 \text{ (mA cm}^{-2}\text{)}$, 630(mV) and (0.7) respectively and increase the efficiency of the Si solar cell 13.8 % instead of 10.3% conversion efficiency before the deposited.



Fig.(1) Experimental setup.



Fig.(2) Freshly prepared colloidal Titanium dioxide NPs in water (white color).

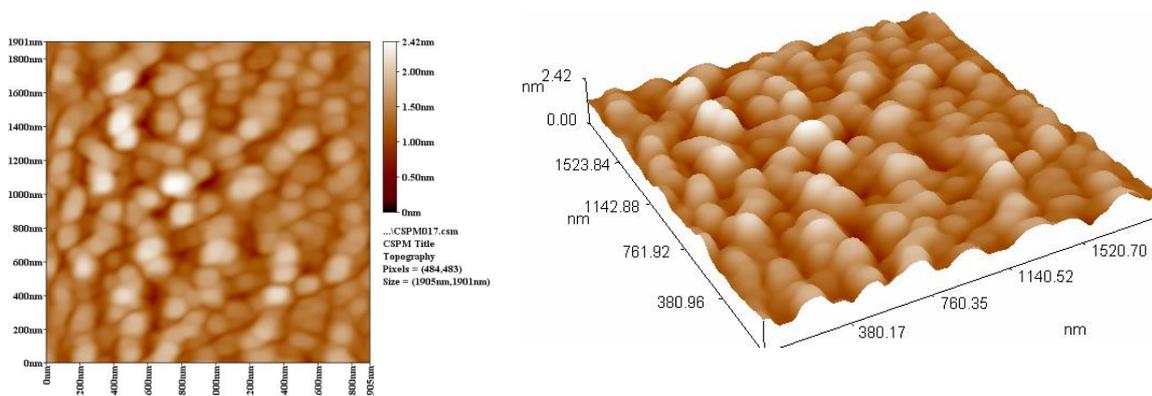


Fig.(3) AFM images of TiO_2 NPs ablated in water.

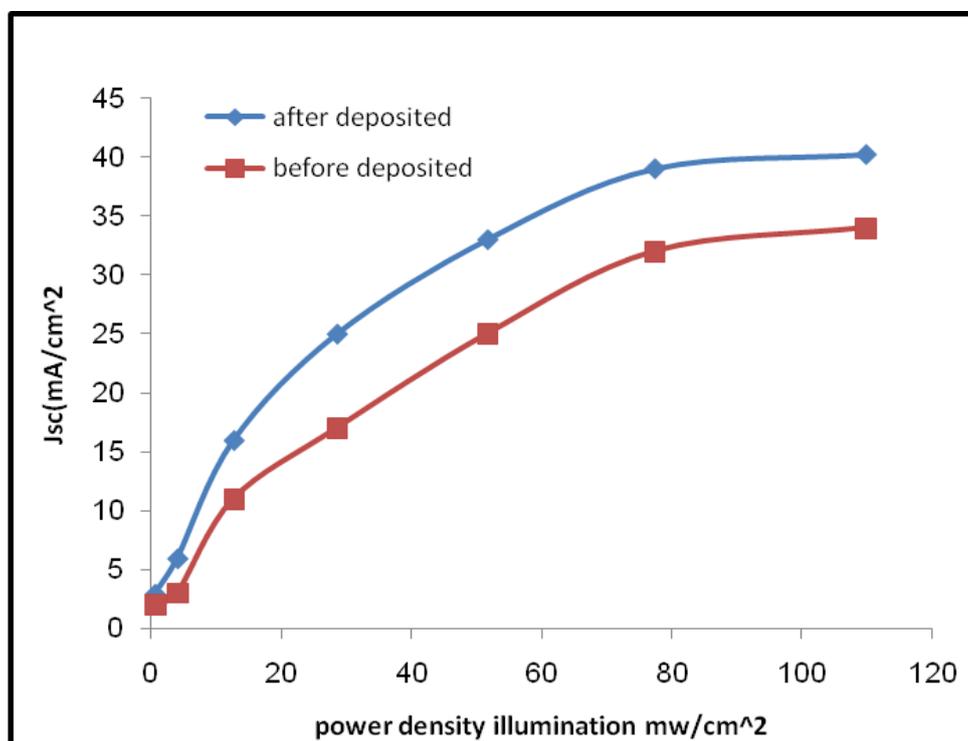


Fig.(4) Current density J_{sc} as a function of illumination plot after and before deposition.

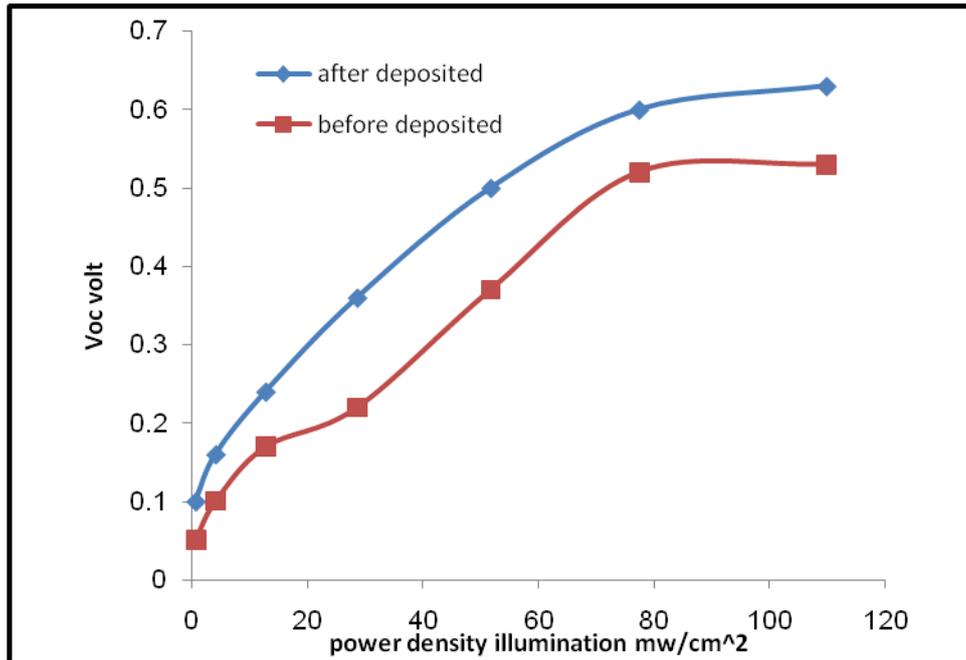


Fig. (5) Open circuit voltage V_{oc} as a function of illumination plot after and before deposition.

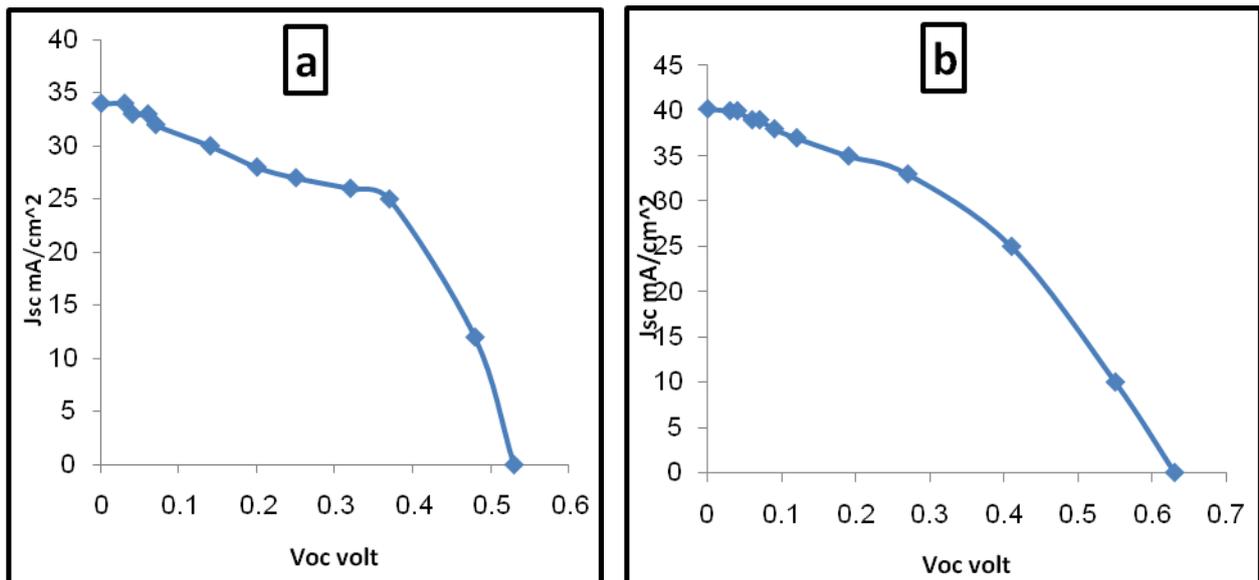


Fig.(6) The Photovoltaic Performance a) before deposited, b) after deposited Titaniumoxide NPs.

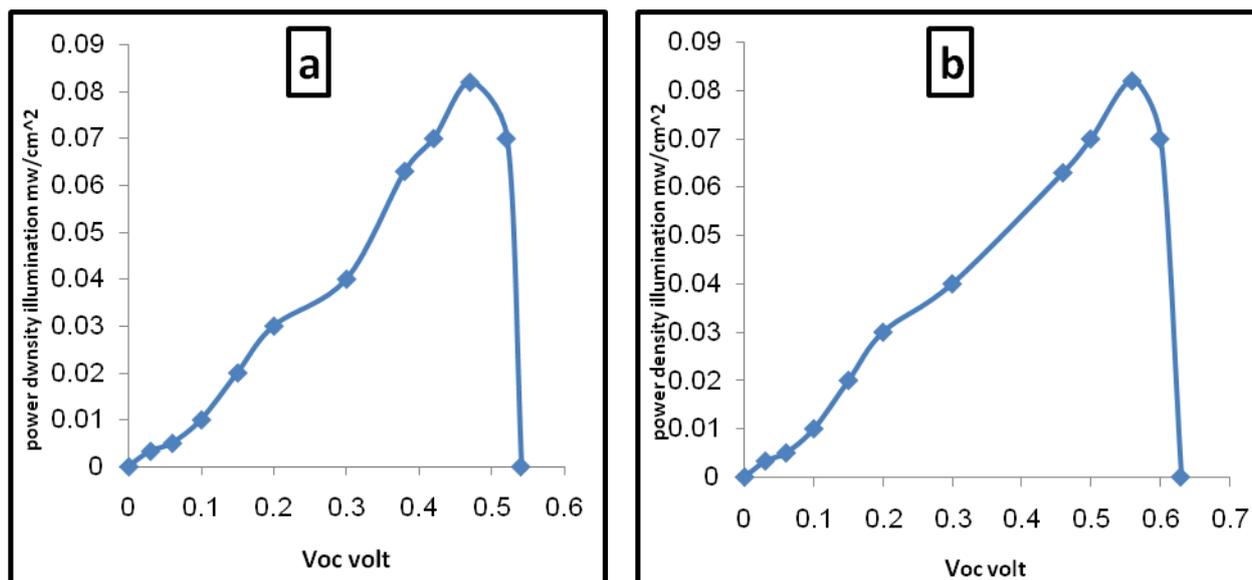


Fig.(7) The Output power a) before deposited, b) after deposited Titaniumoxide NPs.

References

- [1] Zhou Y, Ma R Z, Ebina Y, Takada K, Sasaki T. Multilayer Hybrid;"Films Of Titania Semiconductor Nanosheet And Silver Metal Fabricated Via Layer-By-Layer Self-Assembly And Subsequent UvIrradiation"; J. Chem Mater, 18: 1235–1239, 2010.
- [2] Tang H, Berger H, Schmid P E, Le Vy F;" Photoluminescence In Tio2 Anatase Single Crystals"; J. Solid State Commune, 87: 847–850, 1993.
- [3] Kuang D, Brillet J, Chen P, Takata M, Uchida S, Miura H, Sumioka K, Zakeeruddin S M, Gratzel M;"Application Of Highly Ordered Tio2 Nanotube Arrays In Flexible Dye-Sensitized Solar Cells"; J. Acs Nano, 2: 1113–1116, 2011.
- [4] Hosono E, Fujihara S, Imai H, Honma I, Masaki I, Zhou H S;" One-Step Synthesis Of Nano-Micro Chestnut Tio2 With Rutile Nanopins On The Microanatase Octahedron"; J. Acs Nano, 1: 273–278, 2012.
- [5] Fox M A, Dulay M T;" Heterogeneous Photo Catalysis " ;J. Chemrev, 93: 341–357, 1993.
- [6] Dhage S R, Choube V D, Samuel V, Ravi V;" Synthesis Of Nanocrystalline Tio2 At 100"; J. Mater Lett, 58: 2310–2313, 2009.
- [7] Hu Y, Tsai H L, Huang C L;"Effect Of Brookite Phase On The Anatase-Rutile Transition In Titania Nanoparticles"; J. J Eur Ceram Soc, 23: 691–696, 2003.
- [8] Liu P S, Cai W P, Zeng H B;" Fabrication And Size-Dependent Optical Properties Of Feo Nanoparticles Induced By Laser Ablation In A Liquid Medium"; J. J Physchem C, 112: 3261–3266, 2013.
- [9] Zeng H B, Liu P S, Cai W P;" Aging-Induced Self-Assembly Of Zn/Zno Treelike Nanostructures From Nanoparticles And Enhanced Visible Emission"; J. Cryst Growth Des, 7: 1092–1097, 2011.
- [10] Usui H, Sasaki T, Koshizaki N;" Effect Of Alkyl Chain Length On Layered Structure Of Zn Nanocomposites Prepared By Laser Ablation Of Zn In Aqueous Solution Of Sodium Alkyl Sulfate"; J. Chemlett, 34: 700, 2005.
- [11] Wagner C D, Riggs W M, Davis L E, Moulder J F, Mullenberg G E;"Handbook Of X-Ray Photoelectron Spectroscopy"; M. Minnesota Perkin-Elmer Corporation, 38, 1979.
- [12] Ocal C, Ferrer;" Low Temperature Diffusion Of Pt And Au Atoms Through Thin Tio2 Films On A Ti Substrate"; J. Surface Science, 191: 147–156, 1987.
- [13] Moss T S;" Optical Properties Of Solids"; M. London: Butterworth, 34, 1961.

[14] Kandori K, Kon-No K, Kitahara A;" Formation Of Ionic Water Oil Microemulsions And Their Application In The Preparation Of TiO_2 Particles"; J. J Colloid Interface Sci, 122: 78–82, 2012.

الخلاصة

يهدف هذا البحث الى تحضير محلول للجسيمات النانوية لأكسيد التيتانيوم باستخدام طريقة الاستئصال بالليزر النبضي تم تحضير الجسيمات النانوية لأكسيد التيتانيوم بزمن ٧ نانوثانية بواسطة الليزر نوع نيديميوم ياك لهدف من التيتانيوم عالي النقاوة (99.99 %) مغطس بالماء, ثم ترسيب غشاء اوكسيد التيتانيوم النانوي على سطح خلية شمسية سليكونية وذلك لزيادة كفاءتها. تم دراسة طبيعة السطح لجسيمات الاوكسيد النانوية باستخدام مجهر القوى الذرية حيث بينت تحاليل مجهر القوى الذرية (AFM) ان معدل الحجم الحبيبي لجسيمات الاوكسيد المستئصلة بالماء هي 150 نانو متر وبلغت قيمة خشونة السطح لغشاء اوكسيد التيتانيوم 2.27 نانو متر. درست الخواص الفولطائية للخلية قبل وبعد ترسيب جسيمات اوكسيد التيتانيوم النانوية المتمثلة بتيار دائرة القصر (Jsc), وفولتية الدائرة المفتوحة (V_{oc}), بالاضافة لعامل المليء (FF) لحالة AM1 حيث بلغت قيمتهم بعد الترسيب 40.2 (mA cm⁻²), 630(mV), و 0.7 على التوالي. لقد ازدادت قيمة كفاءة الخلية الشمسية الى 13.8 % بدلا من 10.3 % بعد ترسيب غشاء اوكسيد التيتانيوم النانوي.