

Synthesis and Liquid Crystalline Behavior of New Amides and Esters Containing λ, ν -Thiazole Ring

Nisreen H. Karam*, Jumbad H. Tomma*, Ammar H. AL-Dujaili* and Nasreen R. Jber**

*Department of Chemistry, College of Education Ibn Al- Haitham, University of Baghdad, Baghdad -Iraq.

**Department of Chemistry, College of Science, Al-Nahrain University, Baghdad-Iraq.

Abstract

Synthesis and study liquid crystalline properties of new compounds with terminal groups of amides ([III]_{a-c}, [IV]_{a-c} and [VI]_n), alkoxy series [V]_n or ester with azo linkage ([IX]_{a-c} and [X]_{a-c}) containing thiazole ring. These series were synthesized by many steps starting from ϵ -hydroxyacetophenone or ϵ -aminoacetophenone.

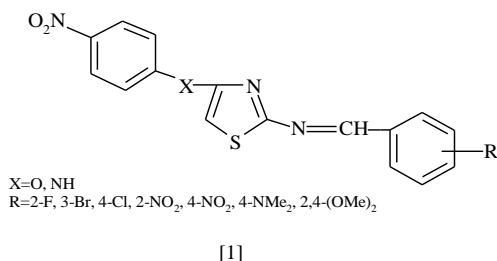
The synthesized compounds were characterized using melting points, FTIR, C.H.N.S analysis and for some of them ¹H NMR spectroscopy. The liquid crystalline properties were studied by hot stage polarizing microscopy and differential scanning calorimetry DSC. All compounds of series [III]_{a-c}, [IV]_{a-c} and compounds [V]_n showed enantiotropic liquid crystal. While the series [VI]_n showed nematic mesomorphism except [VI]_λ did not show any mesomorphic behavior, also the compounds [IX]_{a, b} and [X]_{a, b} showed nematic liquid crystalline, but the compounds [IX]_c and [X]_c did not show any liquid crystalline properties.

Keywords: Thiazole, amides, azo compounds, liquid crystal, heterocyclic.

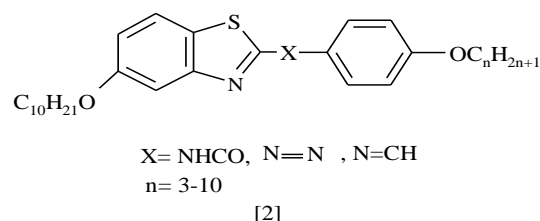
Introduction

Thiazole is very important structural feature in liquid crystals properties^[1]. Some functional groups have proven to be very useful to promote mesomorphic properties. For example, the amide and ester groups are some of the most commonly used, as below.

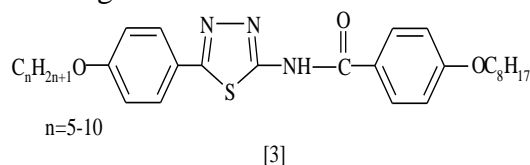
Kuvatov and coworkers, have synthesized and study liquid crystal properties for series containing a thiazole ring [λ] with different terminal substituents (R)^[1].



Belmar and coworker, have synthesized a series of benzothiazole derivatives [ν] of liquid crystals within two different linkage groups (amide and azo) and compared with series of imines^[1]. The compounds with imine and azo linkages showed typical nematic and smectic C mesophases, while the compounds with amide linkage showed a poorer mesomorphism and mainly present a smectic C mesophase.



In addition, synthesized and studied liquid crystalline properties of a new series of amides with λ, ν, ϵ -thiadiazole ring [ν]. The compounds of series [ν] display nematic (N) and smectic C (SC) mesomorphism in the whole range of n studied^[1].



Also, synthesized liquid crystalline amide compounds^[1, v]. Many workers synthesized and study a new series of asymmetric bent-core compounds using azo linkages with equal as well as unequal terminal alkyl chains^[1].

In this study we present the synthesis and mesomorphic behavior of new compounds of thiazole ring with different terminal groups (amide, alkoxy or ester).

Experimental Chemicals

All chemicals were supplied by fluka, merk and Aldrich chemicals Co. with 90% purity at least.

Techniques

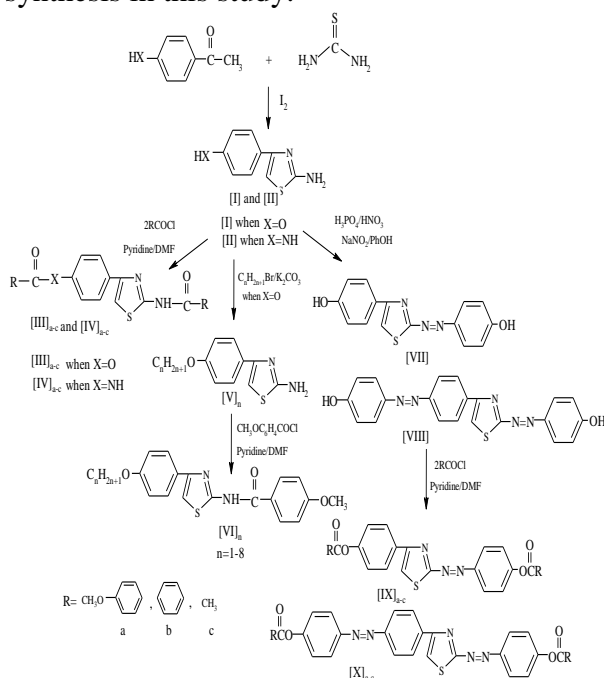
FTIR spectra were recorded on a Shimadzu (Ir prestige-21) using potassium bromide discs. ¹H NMR spectra were carried out on Bruker, ultra-shield 300 MHz Switzerland and are reported in ppm(δ), DMSO and CDCl₃ were used as solvents with TMS as an internal standard. Elemental micro-analyses of some compounds were performed on a (C.H.N.S) : Euro vector, EA 3000A origin: Italy. Amman, Jordan.

Uncorrected melting points were determined using Hot-Stage, Gallen Kamp melting point apparatus.

The transition temperatures and textures of the mesophases were determined using polarized optical microscope model MT993 with hot-stage type INSTEC MK1000 and DSC measurements were conducted with DSC-7, SHIMADZU and TA-7, WS instruments.

Synthesis

The following scheme 1 shows the whole synthesis in this study:



Scheme (1) reaction steps for the synthesis.

Synthesis of 2-Amino 4-(4'-hydroxy or aminophenyl)-1,3-thiazole [I] and [II]

In a 200 mL round bottom flask 2.11g (0.03 mol) iodine was added, thiourea (2.06g, 0.06 mol) and 4-hydroxyacetophenone or 4-aminocetophenone (0.03 mol) were added and the mixture was heated for 6 hrs. on water bath. Cooling, the resulting solid was washed with diethyl ether and a solution of sodium thiosulfate, the precipitate was filtered^[1] and recrystallized from ethanol.

compound [I]; yield 89%, m.p = 148-150 °C
compound [II]; yield 83%, m.p = 110-111 °C.

Synthesis of 2-N-(aryl or methyl amide)-4-[(4'-aryl or methyl carboxylate or amide) phenyl]-1,3-thiazole [III]a-c and [IV]a-c

The compound [I] or [II] (0.01 mol) was dissolved in (1 mL) pyridine and 2 mL DMF in ice bath and acid chloride (0.02 mol) was added, the mixture stirred for 3 hrs at room temperature, afterward added 10% HCl, the precipitate was filtered and washed with water, dried and recrystallized from ethanol.

2-N-(4'-methoxybenzamide)-4-[(4'-methoxybenzoate)phenyl]-1,3-thiazole [III]a yield 80%; FTIR: 3408 (N-H), 2960-2852 (C-H) aliphatic, 1732 (C=O) ester, 1680 (C=O) amide, 1606 (C=C) aromatic, 1161 (C-O) ester.

2-N-benzamide-4-[(4'-benzoate)phenyl]-1,3-thiazole [III]b yield 79%; FTIR: 3396 (N-H), 1750 (C=O) ester, 1670 (C=O) amide, 1090 (C=C) aromatic, 1172 (C-O) ester.

2-N-acetamide-4-[(4'-acetate)phenyl]-1,3-thiazole [III]c yield 80%; FTIR: 3350 (N-H), 2927-2890 (C-H) aliphatic, 1701 (C=O) ester, 1689 (C=O) amide, 1098 (C=C) aromatic, 1166 (C-O) ester.

2-N-(4'-methoxybenzamide)-4-[(4'-methoxybenzamide)phenyl]-1,3-thiazole [IV]a yield 80%; FTIR: 3415 (N-H), 2972-2850 (C-H) aliphatic, 1680, 1660 (C=O) amide, 1606 (C=C) aromatic.

2-N-benzamide-4-[(4'-benzamide)phenyl]-1,3-thiazole [IV]b

yield 79%; FTIR: 3342, 1670, 1603 (C=O)
amide, 1600 (C=C) aromatic

2-N-acetamide-4-(4'-acetamide)phenyl-1,3-thiazole [IV]; yield 70%; FTIR: 3410 (N-H), 2930-2800 (C-H) aliphatic, 1680, 1603 (C=O) amide, 1097 (C=C) aromatic.

Synthesis of 2-Amino 4-(4'-n-alkoxyphenyl)-1,3-thiazole [V]_n

The compound [I] (0.28g, 0.0010 mol) and anhydrous potassium carbonate (0.8g, 0.012 mol) were dissolved in acetone 5 mL, next n-alkyl bromide (0.004 mol) was added, the mixture was refluxed overnight and poured into ice water. The precipitate filtered and washed with water^[11], the solid residue was dried and recrystallized from ethanol.

2-amino 4-(4'-methoxyphenyl)-1,3-thiazole [V]₁

Yield 60%; FTIR: 3441, 3273, 3118 (NH_r), (N-H), 2997-2837 (C-H) aliphatic, 1620 (C=N) endocyclic, 1606 (C=C) aromatic, 1292 (C-O) ether.

2-amino 4-(4'-ethoxyphenyl)-1,3-thiazole [V]₂

Yield 60%; FTIR: 3446, 3270, 3116 (NH_r), (N-H), 2980-2877 (C-H) aliphatic, 1629 (C=N) endocyclic, 1606 (C=C) aromatic, 1288 (C-O) ether.

2-amino 4-(4'-n-propoxyphenyl)-1,3-thiazole [V]₃

Yield 83%; FTIR: 3402, 3288, 3170 (NH_r), (N-H), 2970-2872 (C-H) aliphatic, 1620 (C=N) endocyclic, 1606 (C=C) aromatic, 1286 (C-O) ether.

2-amino 4-(4'-n-butoxyphenyl)-1,3-thiazole [V]₄

yield 70%; FTIR: 3421, 3292, 3167 (NH_r), (N-H), 2904-2870 (C-H) aliphatic, 1633 (C=N) endocyclic, 1608 (C=C) aromatic, 1286 (C-O) ether.

2-amino 4-(4'-n-pentoxyphenyl)-1,3-thiazole [V]₅

Yield 74%; FTIR: 3383, 3311, 3124 (NH_r), (N-H), 2904-2860 (C-H) aliphatic, 1639 (C=N) endocyclic, 1606 (C=C) aromatic, 1288 (C-O) ether.

2-amino 4-(4'-n-hexoxy phenyl)-1,3-thiazole [V]₆

Yield 77%; FTIR: 3448, 3296, 3100 (NH_r), (N-H), 2929-2806 (C-H) aliphatic, 1633 (C=N) endocyclic, 1606 (C=C) aromatic, 1294 (C-O) ether.

2-amino 4-(4'-n-heptoxy phenyl)-1,3-thiazole [V]₇

Yield 83%; FTIR: 3317, 3180, 3124 (NH_r), (N-H), 2904-2804 (C-H) aliphatic, 1641 (C=N) endocyclic, 1606 (C=C) aromatic, 1288 (C-O) ether.

2-amino 4-(4'-n-octoxy phenyl)-1,3-thiazole [V]₈

Yield 78%; FTIR: 3412, 3300, 3124 (NH_r), (N-H), 2903-2804 (C-H) aliphatic, 1631 (C=N) endocyclic, 1606 (C=C) aromatic, 1286 (C-O) ether.

Synthesis of 2-N-(4'-Methoxybenzamide)-4-(4'-n-alkoxy phenyl)-1,3-thiazole [VI]_n

These compounds were synthesized in a similar way for synthesized compound [III]_{a-c}, except using compounds [V]_n instead of compound [I] and used 0.001 mol instead of 0.002 mol from anisoyl chloride. the product was filtered and recrystallized from ethanol to give new product.

2-N-(4'-methoxybenzamide)-4-(4'-methoxyphenyl)-1,3-thiazole [VI]₁

yield 70%; FTIR: 3414 (N-H), 3070 (C-H) aromatic, 2980-2846 (C-H) aliphatic, 1688 (C=O) amide, 1620 (C=N) endocyclic, 1608 (C=C) aromatic.

2-N-(4'-methoxybenzamide)-4-(4'-ethoxyphenyl)-1,3-thiazole [VI]₂

Yield 78%, FTIR: 3446 (N-H), 3071 (C-H) aromatic, 2960-2840 (C-H) aliphatic, 1680 (C=O) amide, 1628 endocyclic, 1608 (C=C) aromatic.

2-N-(4'-methoxybenzamide)-4-(4'-n-propoxyphenyl)-1,3-thiazole [VI]₃

Yield 80%; FTIR: 3298 (N-H), 3070 (C-H) aromatic, 2980-2846 (C-H) aliphatic, 1687 (C=O) amide, 1620 endocyclic, 1608 (C=C) aromatic.

2-N-(4'-methoxybenzamide)-4-(4'-n-butoxyphenyl)-1,3-thiazole [VI]₄

Yield 77%; FTIR: 3410 (N-H), 3065 (C-H) aromatic, 2908-2840 (C-H) aliphatic, 1697 (C=O) amide, 1637 endocyclic, 1608 (C=C) aromatic.

2-N-(4-methoxybenzamide)-4-(4-n-pentoxyphenyl)-1,3-thiazole [VI]•

Yield 80%; FTIR: 3415 (N-H), 3067 (C-H) aromatic, 2960-2846 (C-H) aliphatic, 1660 (C=O) amide, 1630, 1608 (C=C) aromatic.

2-N-(4-methoxybenzamide)-4-(4-n-hexoxyphenyl)-1,3-thiazole [VI]•

Yield 79%; FTIR: 3300 (N-H), 3060 (C-H) aromatic, 2900-2847 (C-H) aliphatic, 1660 (C=O) amide, 1630 endocyclic, 1608 (C=C) aromatic.

2-N-(4-methoxybenzamide)-4-(4-n-heptoxyphenyl)-1,3-thiazole [VI]•

Yield 77%; FTIR: 3383 (N-H), 3061 (C-H) aromatic, 2908-2846 (C-H) aliphatic, 1662 (C=O) amide, 1632 endocyclic, 1608 (C=C) aromatic.

2-N-(4-methoxybenzamide)-4-(4-n-octoxyphenyl)-1,3-thiazole [VI]•

Yield 83%; FTIR: 3419 (N-H), 3061 (C-H) aromatic, 2960-2805 (C-H) aliphatic, 1693 (C=O) amide, 1643 (C=N) endocyclic, 1608 (C=C) aromatic.

Synthesis of 2-(4-Hydroxyphenyl azo)-4-(4-hydroxy phenyl)-1,3-thiazole [VII] and 2-(4-Hydroxyphenyl azo)-4-(4-hydroxy phenyl azo) phenyl]-1,3-thiazole [VIII]

The azo compounds [VII] and [VIII] were synthesized according to literature^[11].

Compound[VII]; yield 70%, m.p = 128-130°C.

Compound[VIII]; yield 70%, m.p > 290°C.

Synthesis of 2-(4-aryl or methyl carboxylate phenyl azo)-4-(4-aryl or methyl carboxylate phenyl)-1,3-thiazole [IX]_{a-c} and 2-(4-Aryl or methyl carboxylate phenyl azo)-4-(4-aryl or methyl carboxylate phenyl azo) phenyl]-1,3-thiazole[X]_{a-c}

These compounds were synthesized in a similar way for synthesized compound [III]_{a-c}, except using the compounds [VII] and [VIII], respectively instead of compound [I], the

product was filtered and recrystallized from ethanol to give new product.

2-[4-(4-methoxy benzoate) phenyl azo]-4-[4-(4-methoxy benzoate) phenyl]-1,3-thiazole [IX]_a

Yield 79%; FTIR: 2996-2840 (C-H) aliphatic, 1730, 1680 (C=O) ester, 1605 (C=C) aromatic, 1050 (N=N), 1222 (C-O) ester.

2-(4-benzoatephenylazo)-4-(4-benzoatephenyl)-1,3-thiazole [IX]_b

Yield 70%; FTIR: 1759, 1725 (C=O) ester, 1098 (C=C) aromatic, 1030 (N=N), 1213 (C-O) ester.

2-(4-acetatephenylazo)-4-(4-acetatephenyl)-1,3-thiazole [IX]_c

Yield 70%; FTIR: 2926-2887 (C-H) aliphatic, 1726, 1670 (C=O) ester, 1080 (C=C) aromatic, 1030 (N=N), 1185 (C-O) ester.

2-[4-(4-methoxy benzoate) phenyl azo]-4-[4-(4-methoxy benzoate phenyl azo)phenyl]-1,3-thiazole[X]_a

Yield 77%; FTIR: 2937-2841 (C-H) aliphatic, 1739, 1715 (C=O) ester, 1605 (C=C) aromatic, 1033 (N=N), 1222 (C-O) ester.

2-(4-benzoatephenylazo)-4-[4-(4-benzoatephenylazo) phenyl]-1,3-thiazole[X]_b

Yield 83%; FTIR: 1753, 1701 (C=O) ester, 1098 (C=C) aromatic, 1029 (N=N), 1209 (C-O) ester.

2-(4-acetatephenyl azo)-4-[4-(4-acetate phenyl azo) phenyl]-1,3-thiazole[X]_c

Yield 78%; FTIR: 2943-2881 (C-H) aliphatic, 1701, 1720 (C=O) ester, 1605 (C=C) aromatic, 1033 (N=N), 1219 (C-O) ester.

Results and Discussion

The characteristic FTIR spectra, showed the disappearance of absorption stretching bands of carbonyl group C=O and C=S of starting materials with the appearance of a stretching bands in region (3500-3199) cm⁻¹ and (3305-3182) cm⁻¹ respectively due to NH, NH₂ groups and appearance a new stretching band at 1622 cm⁻¹ and 1625 cm⁻¹

due to C=N for thiazole ring moiety, respectively.

The ^1H NMR spectrum of compound [I] showed the following signals: a singlet signal at δ 9.03 ppm for one proton could be attributed of OH group, two doublet pair in the region δ (7.70-7.82) ppm due to four aromatic protons, a broad signal at δ 7.07 ppm for two protons of NH_ν group. The signal at δ 7.74 ppm for one proton of thiazole ring^[17].

The FTIR spectra of compounds [III]_{a-c} (ester and amide) showed a significant bands in the region (1732-1701) cm^{-1} and (1680-1689) cm^{-1} which could be attributed to stretching vibration band of the carbonyl of ester and amide group, respectively showed disappearance of stretching absorption bands due to O-H and NH_ν of compound [I]. While The FTIR spectra of amides [IV]_{a-c} showed a significant band in the region (1680-1603) cm^{-1} due to carbonyl of amide group, also showed the disappearance of stretching absorption band due to NH_ν group of compound [I].

The ^1H NMR spectrum of compound [III]_a showed: a singlet signal at δ 12.5 ppm for proton of NH group, two doublet of doublets in the region δ (7.13-8.08) ppm could be attributed to twelve aromatic protons of three benzene rings, a weak signal at δ (7.30) ppm for one proton of thiazole ring, two sharp signals at δ (3.88) ppm and (3.82) ppm for six protons of two OCH_ν groups.

While the ^1H NMR spectrum for amide compound [IV]_b showed: signal at δ 10.5 ppm for two protons of two NH groups, multiplet signal in the region δ (7.57-8.12) ppm could be attributed to twelve aromatic protons of three benzene rings and signal at δ (7.01) ppm for proton of a thiazole ring.

The FTIR spectra showed the disappearance of the stretching band of OH group and appearance of new absorption stretching bands of the aliphatic C-H group in the region (2997-2804) cm^{-1} for alkyl groups.

Elemental analysis result of compound [V]_c is good agreement with the theoretical data Elemental analysis of compound [V]_c :

Calc.: C% = 70.11, H% = 7.97, N% = 10.80, S% = 12.40

Found: C% = 74.84, H% = 7.64, N% = 10.00, S% = 12.61

The FTIR spectra showed disappearance bands of NH_ν , NH_ν groups of compounds [V]_n with appearance of a new stretching band of NH group in range (3414-3419) cm^{-1} , also appearance carbonyl amide ($\text{NHC}=\text{O}$) in range (1688-1693) cm^{-1} .

The ^1H NMR spectrum for compound [VI]_c showed: singlet signal at δ 8.0 ppm for one proton of NH group, two doublet in region δ (7.93-8.11) ppm could be attributed for eight aromatic protons, a singlet signal at δ (7.88) ppm could be attributed to proton of thiazole ring, a triplet signal at δ (3.87-3.89) ppm duo to protons of OCH_ν and OCH_ν terminal groups, a multiplet signal at δ (1.40-1.82) ppm for six proton of three groups of $-\text{CH}_\nu-$ and a triplet signal at δ (0.93-0.97) ppm due to three protons of CH_ν group.

The azo compounds [VII] and [VIII] were synthesized by coupling between diazonium salt of the amino thiazole [I] or [II], respectively with phenol at -5°C . The structure of these compounds were studied by melting points, FTIR and C.H.N.S. analysis. The characteristic FTIR absorption bands showed the disappearance of two absorption bands due to NH_ν stretching of compound [I] and [II] together with the appearance of a stretching broad band around 3292 cm^{-1} and (3404-3440) cm^{-1} due to the intermolecular hydrogen bonding of O-H bond, respectively. It also shows a band at 1021 cm^{-1} and 1016 cm^{-1} due to a stretching band of $\text{N}=\text{N}$ group, respectively.

Elemental analysis of compound [VII] :

Calc.: C% = 70.70, H% = 3.70, N% = 14.14, S% = 10.77

Found: C% = 70.22, H% = 3.90, N% = 14.22, S% = 10.66

These results are good agreement with proposed structure.

The compounds [IX]_{a-c} and [X]_{a-c} were synthesized by reaction of one mole of compounds [VII] or [VIII] with two moles of different acid chloride, respectively in presence of dry pyridine as acceptor. The structures of these compounds were studied by melting points, FTIR and ^1H NMR spectroscopy.

The FTIR spectra showed a significant bands in the region (1730-1670) cm^{-1} and (1739-1720) cm^{-1} which could be attributed to

stretching vibration of the carbonyl of ester groups, respectively. They also showed the disappearance of absorption band due to ν O-H of compounds [VII] and [VIII].

The $^1\text{H NMR}$ spectrum for compound [X]_a showed: two doublet in region $\delta(7.0-8.5, 8.5-9.0)$ ppm could be attributed for aromatic protons of benzene rings and the proton of thiazole ring (which is interference with the aromatic protons of benzene), singlet signal at $\delta(3.7, 3.8)$ ppm for six protons of two OCH_3 groups.

Mesomorphic properties

The transition temperatures of compounds [III]_{a-c} and [IV]_{a-c} were summarized in Table (1). When compared between two above series, found that compounds [III]_a and [III]_b showed both smectic and nematic phases, while the compound [III]_c and all of compounds [IV]_{a-c} showed only nematic phase, this explain that compounds [III]_a and [III]_b favored a layer smectic order because was fully compatible with a molecular arrangement resulting from the intermolecular H-bonding which develops in the tilt direction and thus stabilizes the tilt of the smectic phase^[7] (as shown in Fig.(1)) in addition these compounds containing three phenyl rings for this reasons, displayed a tilted smectic C mesophase in addition nematic phase.

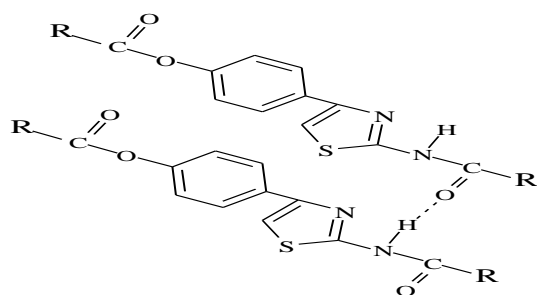
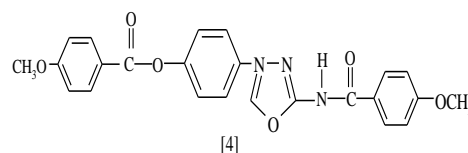


Fig.(1) Hydrogen bonding formation for compounds [III]_{a,b}.

But the compound [III]_c containing one phenyl ring and two methyl groups (shorter chain) these decrease in lateral attraction forces and not appears smectic phase. While compounds [IV]_{a-c} did not have ester group and low lateral forces (ratio(t/l) is high) for this reason, the compounds [IV]_{a-c} showed a nematogenic behaviors.

When compared compound [III]_a in this work with compound [ξ] in another study^[17], found the compound [ξ] with oxadiazole ring

did not show any liquid crystal mesophase, but compound [III]_a with thiazole ring in this study showed Sc and nematic phase.



While the series [V]_n with $n=1-6$ showed only nematic phases but when $n=7-8$ displayed nematic phases and smectic C, as shown in Fig.(7) for DSC thermogram of compound [V]₇ and Fig.(8) for smectic C (SmC) texture of compound [V]₈. This behavior can be described to the chain length of alkoxy end group that interferes with liquid crystalline ordering, this could be explain in term of terminal/lateral (t/l) interaction forces ratio. When this ratio is high, compounds tend to give less ordered mesophases (nematic mesophase), While this ratio is low, the compounds tend to give high order mesophases (smectic phases), the transition temperatures of series [V]_n are given in Table (9). All compounds of series [VI]_n showed nematic phase, while the compound [VI]₈ failed in the formation of liquid crystalline phase, the transition temperature are given in Table (9). A plot of transition temperatures against the number (n) of carbon atoms in the lateral alkoxy chain are shown in Fig.(ξ). The effects of the terminal chain length on the transition temperatures and phases behavior observed in this series are in accordance with those observed for conventional low molar mass mesogens.

All compounds [IX]_{a,b} and [X]_{a,b} showed only nematic texture, as shown in Figure 9 for compound [IX]_b but the compounds [IX]_c and [X]_c failed in the formation of liquid crystalline phase. In this case we can explain that the end groups polarizability (acetyl groups) are not large enough to give enantiotropic nematic phase, therefore these compounds did not show appear any liquid crystal phases. The transition temperatures are given in Table (ξ).

Table (1)
Phase transition of compounds [III]_{a-c} and [IV]_{a-c}.

Compound	Phase transition
[III] _a	Cr $\xrightleftharpoons{78}$ Sc $\xrightleftharpoons{95}$ N $\xrightleftharpoons{144}$ I
[III] _b	Cr $\xrightleftharpoons{95}$ Sc $\xrightleftharpoons{120}$ N $\xrightleftharpoons{162}$ I
[III] _c	Cr $\xrightleftharpoons{87}$ N $\xrightleftharpoons{99}$ I
[IV] _a	Cr $\xrightleftharpoons{60}$ N $\xrightleftharpoons{91}$ I
[IV] _b	Cr $\xrightleftharpoons{118}$ N $\xrightleftharpoons{147}$ I
[IV] _c	Cr $\xrightleftharpoons{65}$ N $\xrightleftharpoons{100}$ I

Table (2)
Phase transition of series [V]_n.

Compound	Phase transition
[V] ₁	Cr $\xrightleftharpoons{155}$ N $\xrightleftharpoons{171}$ I
[V] ₂	Cr $\xrightleftharpoons{235}$ N $\xrightleftharpoons{250}$ I
[V] ₃	Cr $\xrightleftharpoons{132}$ N $\xrightleftharpoons{151}$ I
[V] ₄	Cr $\xrightleftharpoons{73}$ N $\xrightleftharpoons{119}$ I
[V] ₅	Cr $\xrightleftharpoons{109}$ N $\xrightleftharpoons{158}$ I
[V] ₆	Cr $\xrightleftharpoons{74}$ Sc $\xrightleftharpoons{101}$ N $\xrightleftharpoons{185}$ I
[V] ₇	Cr $\xrightleftharpoons{54}$ Sc $\xrightleftharpoons{70}$ N $\xrightleftharpoons{102}$ I
[V] ₈	Cr $\xrightleftharpoons{49}$ Sc $\xrightleftharpoons{85}$ N $\xrightleftharpoons{110}$ I

Table (3)
Phase transition of series [VI]_n.

Compound	Phase transition
[VI] ₁	Cr $\xrightleftharpoons{85}$ N $\xrightleftharpoons{122}$ I
[VI] ₂	Cr $\xrightleftharpoons{90}$ N $\xrightleftharpoons{120}$ I
[VI] ₃	Cr $\xrightleftharpoons{83}$ N $\xrightleftharpoons{117}$ I
[VI] ₄	Cr $\xrightleftharpoons{85}$ N $\xrightleftharpoons{105}$ I
[VI] ₅	Cr $\xrightleftharpoons{79}$ N $\xrightleftharpoons{102}$ I
[VI] ₆	Cr $\xrightleftharpoons{83}$ N $\xrightleftharpoons{99}$ I
[VI] ₇	Cr $\xrightleftharpoons{76}$ N $\xrightleftharpoons{91}$ I
[VI] ₈	Cr $\xrightleftharpoons{71-72}$ I

Table (4)
Phase transition of compounds [IX]_{a-c} and [X]_{a-c}.

Compound	Phase transition
[IX] _a	Cr $\xrightleftharpoons{85}$ N $\xrightleftharpoons{117}$ I
[IX] _b	Cr $\xrightleftharpoons{45}$ N $\xrightleftharpoons{118}$ I
[IX] _c	Cr $\xrightleftharpoons{144-145}$ I
[X] _a	Cr $\xrightleftharpoons{47}$ N $\xrightleftharpoons{89}$ I
[X] _b	Cr $\xrightleftharpoons{129}$ N $\xrightleftharpoons{231}$ I
[X] _c	Cr $\xrightleftharpoons{> 295}$ I

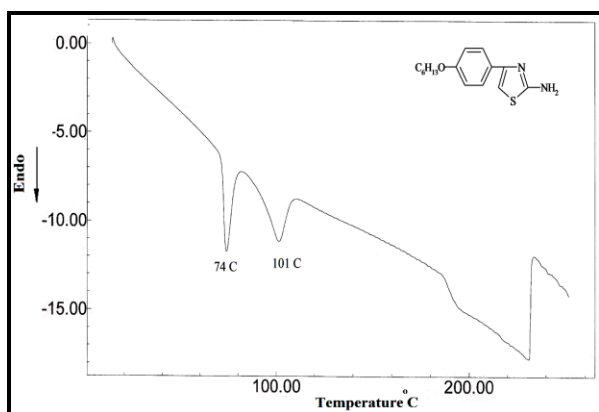


Fig. (ϕ) DSC Thermogram of compound [V].

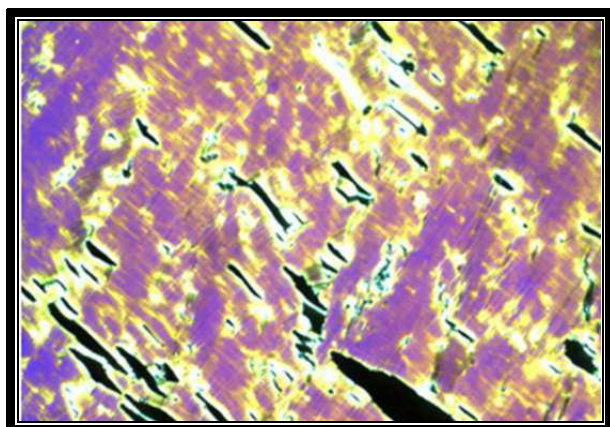


Fig. (ϕ) Smectic C (SmC) texture of compound [V].

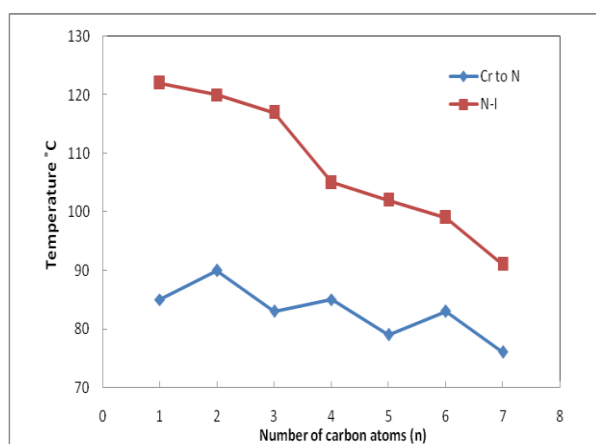


Fig. (ζ) Dependence of transition temperatures on the increasing number of carbon atoms (n) in the terminal alkoxy chains for the compounds of series [VI]_n.

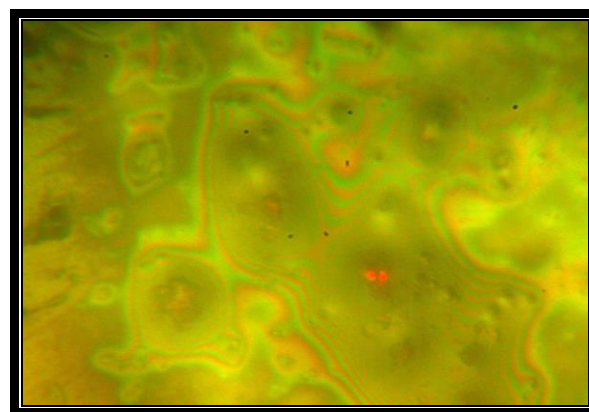


Fig. (ϑ) Nematic texture of the compound [IX]_b.

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تناول هذا البحث تحضير ودراسة الخواص البلورية السائلة لمركبات جديدة تحتوي على مجاميع طرفية أميدية [V]_n, [IV]_{a-c}, [III]_{a-c}, [VI]_n, سلسلة الكوكسي [X]_{a-c} و [IX]_{a-c} تحتوي على حلقة الثيازول. حضرت هذه السلاسل بعدد من الخطوات تبدأ بـ 4- هيدروكسي أسيتوفينون أو 4- أمينو أسيتوفينون. شخّصت التراكيب الكيميائية للمركبات المحضرة بواسطة درجات الانصهار FTIR، تحليل C.H.N.S. ولبعض منها طيف ¹H NMR، أما الخواص البلورية السائلة فتم دراستها بواسطة مجهر الضوء المستقطب المزود بمنصة تسخين ومسعر المسح التفاضلي DSC. جميع مركبات السلاسل [V]_n و [IV]_{a-c}, [III]_{a-c} أظهرت خواص بلورية سائلة أما مركبات السلسلة [VI]_n فأظهرت طوراً نيماتياً ما عدا المركب [VI]₈ فأنة لم يظهر أي خواص بلورية سائلة. بالإضافة إلى ذلك المركبات [IX]_{a,b} و [X]_{a,b} أظهرت الطور النيماتى لكن المركبين [IX]_c و [X]_c لم يظهر أي خواص بلورية سائلة.